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## Peter Norris and Matthias Zeller*

Department of Chemistry, Youngstown State University, 1 University Plaza, Youngstown, OH 44555-3663, USA

Correspondence e-mail: mzeller@cc.ysu.edu

## Key indicators

Single-crystal X-ray study
$T=200 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.039$
$w R$ factor $=0.110$
Data-to-parameter ratio $=10.3$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 2,3,4-Tri-O-acetyl- $\beta$-L-fucopyranosyl azide

The title compound, $\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{7}$, crystallizes with two crystallographically independent molecules in the asymmetric unit. Both virtually identical molecules exhibit the expected chair conformation.

## Comment

2,3,4-Tri- $O$-acetyl- $\beta$-L-fucopyranosyl azide, (2), is formed as a result of an $\mathrm{S}_{N} 2$ displacement of bromide by the azide anion from the corresponding $\alpha$-fucopyranosyl bromide, (1) (Cao \& Roy, 1996), or by the Lewis acid-promoted reaction of $1,2,3,4-$ tetra- $O$-acetyl-L-fucose with $\mathrm{TMSN}_{3}$ (Györgydeák \& Szilágyi, 1985). Azide (2) has been employed in Strecker and Ugi chemistry (Kunz et al., 1991) and as a precursor to peptidosaccharide derivatives (Subramaniam, 1997).


The solid-state structure of (2) was determined by X-ray diffraction at 100 K (Fig. 1). It crystallizes in the non-centrosymmetric space group $P 2_{1}$ with $Z=4$ and two crystallographically independent molecules in the asymmetic unit. Only marginal differences are found between the two molecules, the weighted r.m.s. deviation being $0.1016 \AA$. The derived bond lengths and angles are within the expected ranges, and the six-membered rings exhibit the chair conformation expected for a fucopyranose. The $O$-acetyl groups at


Figure 1
One of the independent molecules of (2), showing $50 \%$ probability displacement ellipsoids.

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## organic papers

atom C 2 , and C 22 for the second molecule, are located in axial positions, and all other non- H substituents are found in equatorial positions. The acetate groups, as well as the azide group of each molecule, are approximately parallel to each other, as may be seen from Fig. 1.

The $\beta$-configuration of (2) with the azide group occupying the equatorial position at the anomeric C atom is clearly confirmed. The azide functional group is close to linear [ $\mathrm{N} 1-$ $\mathrm{N} 2-\mathrm{N} 3=172.2(3)^{\circ}$ and $\left.\mathrm{N} 4-\mathrm{N} 5-\mathrm{N} 6=171.7(3)^{\circ}\right]$ and the $\mathrm{C} 5-\mathrm{N} 1-\mathrm{N} 2$ and $\mathrm{C} 25-\mathrm{N} 4-\mathrm{N} 5$ angles are 114.1 (2) and 113.3 (2) ${ }^{\circ}$, respectively. The $\mathrm{N} \equiv \mathrm{N}$ triple-bond distances $\mathrm{N} 2 \equiv \mathrm{~N} 3$ and $\mathrm{N} 5 \equiv \mathrm{~N} 6$ are 1.120 (3) and 1.112 (3) $\AA$, respectively, and the $\mathrm{N}-\mathrm{N}$ single bond lengths $\mathrm{N} 1-\mathrm{N} 2$ and $\mathrm{N} 4-\mathrm{N} 5$ are 1.230 (3) and 1.246 (3) $\AA$, respectively.

## Experimental

The sample of (2) was prepared according to the method of Cao \& Roy (1996). Crystals suitable for single-crystal X-ray diffraction were obtained by slow cooling of a hot ethanol solution.

## Crystal data

```
\(\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{~N}_{3} \mathrm{O}_{7}\)
\(M_{r}=315.29\)
Monoclinic, \(P 2_{1}\)
\(a=10.8447\) (6) A
\(b=8.3527\) (5) \(\AA\)
\(c=17.443\) (1) A
\(\beta=96.257\) (1) \({ }^{\circ}\)
\(V=1570.62(16) \AA^{3}\)
\(Z=4\)
```

$$
\begin{aligned}
& D_{x}=1.333 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 6342 \\
& \quad \text { reflections } \\
& \theta=2.7-30.6^{\circ} \\
& \mu=0.11 \mathrm{~mm}^{-1} \\
& T=200(2) \mathrm{K} \\
& \text { Block, colourless } \\
& 0.46 \times 0.42 \times 0.40 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Bruker SMART APEX CCD diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS in SAINT-Plus; Bruker, 2003) $T_{\text {min }}=0.868, T_{\text {max }}=0.96$ 16017 measured reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& \begin{array}{c}
w=1 /\left[\sigma^{2}\left(F_{o}^{2}\right)+(0.0699 P)^{2}\right. \\
\quad+0.1865 P] \\
\text { where } P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3 \\
(\Delta / \sigma)_{\max }<0.001 \\
\Delta \rho_{\max }=0.34 \mathrm{e}^{-3} \\
\Delta \rho_{\min }= \\
=
\end{array}{ }^{-0.27 \mathrm{e}^{-3}}
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.039$
$w R\left(F^{2}\right)=0.110$
$S=1.05$
4157 reflections
405 parameters
H -atom parameters constrained

All H atoms were placed in calculated positions and refined with a riding model; the $\mathrm{C}-\mathrm{H}$ distances were $0.98-1.00 \AA$, with $U_{\text {iso }}(\mathrm{H})=$ $1.5 U_{\text {eq }}\left(\mathrm{C}_{\text {methyl }}\right)$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}\left(\mathrm{C}_{\text {methine }}\right)$. Friedel pairs were merged before refinement. The absolute configuration assignment is based on the known configuration of the C atoms, assuming retention of their configuration during the synthesis of the azide. The s.u. values of the cell parameters are taken from the software, recognizing that the values are unreasonably small (Herbstein, 2000).

Data collection: SMART (Bruker, 2002); cell refinement: SAINTPlus (Bruker, 2003); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXTL (Bruker, 2000); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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